



REFERENCES and SOURCES for HITRAN

(Last updated: 29 April 2014)

The reference 0 (zero) is used for all data surviving from the 1986 HITRAN Database. For further details, refer to: L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, "The HITRAN database: 1986 Edition," *Appl.Opt.* **26**, 4058-4097 (1987).

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H₂O [1] 161, 181, 171, 162, 182, 172

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- 43.** Approximation of Ref. 31 with estimated exchange of vibrational quanta.

CO₂ [2] 626, 636, 628, 627, 638, 637, 828, 827, 838, 837

Positions

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Intensities

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Half-widths (air)

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Half-widths (self)

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Temperature dependence of air-broadened half-width

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Pressure shift (air)

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O₃ [3] 666, 668, 686, 667, 676

Positions

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- 41.** A. Barbe, M.R. De Backer, E. Starikova, S.A. Tashkun, X. Thomas, V.G. Tyuterev, "FTS high resolution spectra of $^{16}\text{O}_3$ in 3500 and 5500 cm^{-1} regions. First example of new theoretical 30hermosp for a polyad of strongly coupled states," *JQSRT* **113**, 829-839 (2012).

Half-widths (air)

- 0.** L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith,

- “The HITRAN database: 1986 Edition,” *Appl.Opt.* **26**, 4058-4097 (1987).
- 1.** R.R. Gamache and L.S. Rothman, “Theoretical N₂-broadened halfwidths of ¹⁶O₃,” *Appl.Opt.* **24**, 1651-1655 (1985) scaled by 0.90 (N₂/air) ratio and increased by 9% as recommended by M.A.H. Smith, NASA Langley Research Center, private communication (1990).
 - 2.** Second order polynomial fit in *J* (used for *J*>35) by C.P. Rinsland, NASA Langley Research Center, private communication (1990).
 - 3.** Average values from Ref. 1 as a function of *J* (*J*≤35) used for lines not in database of Ref. 1.
 - 4.** G. Wagner, M. Birk, F. Schreier, and J.-M. Flaud, “Spectroscopic database for ozone in the fundamental spectral regions,” *J.Geophys.Res.* **D107**, 4626 (2002).
 - 5.** Use of values obtained for the v₃ band from Ref. 4.
 - 6.** Use of values obtained for the v₁/ v₂ bands from Ref. 4.
 - 7.** Third order polynomial fit in *J* for transitions where *J* = *K_a* by I.E. Gordon (2008).
 - 8.** J.-M. Flaud, C. Camy-Peyret, C.P. Rinsland, V. Malathy Devi, M.A.H. Smith, A. Goldman, “Improved line parameters for ozone bands in the 10-μm spectral region,” *Appl. Opt.* **29**, 3667-3671 (1990). The polynomial expression in *J'* from this article has been multiplied by a factor of 1.05 (I. Gordon, private communication, 2008).

Half-widths (self)

- 1.** C.P. Rinsland, J.-M. Flaud, A. Goldman, A. Perrin, C. Camy-Peyret, M.A.H. Smith, V. Malathy Devi, D.Chris Benner, A. Barbe, T.M. Stephen, and F.J. Murcay, “Spectroscopic Parameters for Ozone and Its Isotopes: Current Status, Prospects for Improvement, and the Identification of ¹⁶O¹⁶O¹⁷O and ¹⁶O¹⁷O¹⁶O Lines in Infrared Ground-based and Stratospheric Solar Absorption Spectra,” *JQSRT* **60**, 803-814 (1998).
- 2.** M.A.H. Smith, NASA Langley Research Center, private communication (2004).

Temperature dependence of air-broadened half-width

- 1.** Mean value of R.R. Gamache, “Temperature dependence of N₂-broadened halfwidths of ozone,” *J.Mol.Spectrosc.* **114**, 31-41 (1985).
- 2.** G. Wagner, M. Birk, F. Schreier, and J.-M. Flaud, “Spectroscopic database for ozone in the fundamental spectral regions,” *J.Geophys.Res.* **D107**, 4626 (2002).
- 3.** Use of values obtained for the v₃ band from Ref. 2.
- 4.** Use of values obtained for the v₁/ v₂ bands from Ref. 2.

Pressure shift (air)

- 1.** Mean values of M.A.H. Smith, private communication (2004) based on V. Malathy Devi, D.C. Benner, M.A.H. Smith, and C.P. Rinsland, “Air-broadening and shift coefficients of O₃ lines in the v₂ band and their temperature dependence,” *J.Mol.Spectrosc.* **182**, 221-238 (1997); M.A.H. Smith, V. Malathy Devi, D.C. Benner, and C.P. Rinsland, “Temperature dependence of air-broadening and shift coefficients of O₃ lines in the v₁ band,” *J.Mol.Spectrosc.* **182**, 239-259 (1997); M.A.H. Smith, C.P. Rinsland, V. Malathy Devi, and E.S. Prochaska, “Measurements of pressure broadening and shifts of O₃ lines in the 3-μm region,” *J.Mol.Spectrosc.* **164**, 239-259 (1994); M.A.H. Smith, C.P. Rinsland, V. Malathy Devi, and E.S. Prochaska, “Erratum: Measurements of pressure broadening and shifts of O₃ lines in the 3-μm region” by M.A.H. Smith, C.P. Rinsland, V. Malathy Devi, and E.S. Prochaska,” *J.Mol.Spectrosc.* **165**, 596 (1994).

N₂O [4] 446, 456, 546, 448, 447

Positions

0. L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, "The HITRAN database: 1986 Edition," *Appl.Opt.* **26**, 4058-4097 (1987).
1. G. Guelachvili, *Can.J.Phys.* **60**, 1334 (1982).
2. R.A. Toth, "Line-frequency measurements and analysis of N₂O between 900 and 4700 cm⁻¹," *Appl.Opt.* **30**, 5289-5315 (1991).
3. K. Chance, K.W. Jucks, D.G. Johnson, and W.A. Traub, "The Smithsonian Astrophysical Observatory Database SAO92," *JQSRT* **52**, 447-457 (1994).
4. J.W.C. Johns, Z. Lu, M. Weber, J.M. Sirota, and D.C. Reuter, "Absolute Intensities in the v₂ fundamental of N₂O at 17 μm," *J.Mol.Spectrosc.* **177**, 203-210 (1996).
5. L. Daumont, C. Claveau, M.R Debacher-Barrilly, A. Hamdouni, L. Régalia-Jarlot, J.-L. Teffo, S. Tashkun, and V.I. Perevalov, "Line intensities of ¹⁴N₂¹⁶O: the 10 micrometers region revisited," *JQSRT* **72**, 37-55 (2002).
6. R.A. Toth, "Linelist of N₂O parameters from 500 to 7500 cm⁻¹," see <http://mark4sun.jpl.nasa.gov/n2o.html>.
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Intensities

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1. K. Chance, K.W. Jucks, D.G. Johnson, and W.A. Traub, "The Smithsonian Astrophysical Observatory Database SAO92," *JQSRT* **52**, 447-457 (1994).
2. J.W.C. Johns, Z. Lu, M. Weber, J.M. Sirota, and D.C. Reuter, "Absolute Intensities in the v₂ fundamental of N₂O at 17 μm," *J.Mol.Spectrosc.* **177**, 203-210 (1996).
3. L. Daumont, C. Claveau, M.R Debacher-Barrilly, A. Hamdouni, L. Régalia-Jarlot, J.-L. Teffo, S. Tashkun, and V.I. Perevalov, "Line intensities of ¹⁴N₂¹⁶O: the 10 micrometers region revisited," *JQSRT* **72**, 37-55 (2002).
4. L. Daumont, J. Vander Auwera, J.-L. Teffo, V.I. Perevalov, and S.A. Tashkun, "Line Intensity Measurements in ¹⁴N₂¹⁶O and their Treatment using the Effective Dipole Moment Approach," *J.Mol.Spectrosc.* **208**, 281-291 (2001).
5. R.A. Toth, "Linelist of N₂O parameters from 500 to 7500 cm⁻¹," see <http://mark4sun.jpl.nasa.gov/n2o.html>.

Half-widths (air)

0. L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, "The HITRAN database: 1986 Edition," *Appl.Opt.* **26**, 4058-4097 (1987).
1. K. Chance, K.W. Jucks, D.G. Johnson, and W.A. Traub, "The Smithsonian Astrophysical Observatory Database SAO92," *JQSRT* **52**, 447-457 (1994).
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(1984).

- 3.** Third-order polynomial fit of experimental results based on the three works: N. Lacome, A. Levy, and G. Guelachvili, “Fourier transform measurement of self-, N₂-, and O₂-broadening of N₂O lines: temperature dependence of linewidths,” *Appl. Opt.* **23**, 425-434 (1984); R.A. Toth, “N₂- and air-broadened linewidths and frequency-shifts of N₂O,” *JQSRT* **66**, 285-304 (2000); and V. Nemtchinov, C. Sun, and P. Varanasi, “Measurements of Line Intensities and Line Widths in the v₃-fundamental Band of Nitrous Oxide at Atmospheric Temperatures,” *JQSRT* **83**, 267-284 (2004).

Half-widths (self)

- 1.** Third-order polynomial fit of experimental results based on: R.A. Toth, “Line strengths (900-3600 cm⁻¹), self-broadened linewidths, and frequency shifts (1800-2660 cm⁻¹) of N₂O,” *Appl. Opt.* **32**, 7326-7365 (1993).

Temperature dependence of air-broadened half-width

- 1.** Fixed to a constant value of 0.75 based on the two works: N. Lacome, A. Levy, and G. Guelachvili, “Fourier transform measurement of self-, N₂-, and O₂-broadening of N₂O lines: temperature dependence of linewidths,” *Appl. Opt.* **23**, 425-434 (1984); V. Nemtchinov, C. Sun, and P. Varanasi, “Measurements of Line Intensities and Line Widths in the v₃-fundamental Band of Nitrous Oxide at Atmospheric Temperatures,” *JQSRT* **83**, 267-284 (2004).

Pressure shift (air)

- 1.** R.A. Toth, “Linelist of N₂O parameters from 500 to 7500 cm⁻¹,” see <http://mark4sun.jpl.nasa.gov/n2o.html>.

CO [5] 26, 36, 28, 27, 38, 37

Positions

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2. R. Farrenq, G. Guelachvili, A.J. Sauval, N. Grevesse, and C.B. Farmer, “Improved Dunham Coefficients for CO from Infrared Solar Lines of High Rotational Excitation,” *J.Mol.Spectrosc.* **149**, 375-390 (1991).
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Intensities

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2. D. Goorvitch, “Infrared CO Linelist for the $\text{X}^1\Sigma^+$ State,” *Astrophys.J.Suppl.Ser.* **95**, 535-552 (1994).
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4. Fit to the experimental values of K. Sung and P. Varanasi, “Intensities, collision-broadened half-widths, and collision-induced line shifts in the second overtone band of $^{12}\text{C}^{16}\text{O}$ ” *JQSRT* **83**, 445-458 (2004).
5. H.S.P. Müller, F. Schloder, J. Stutzki, and G. Winnewisser, “The Cologne Database for Molecular Spectroscopy, CDMS: a useful tool for astronomers and spectroscopists,” *J.Mol.Struct.* **742**, 215-227 (2005). Data adopted in June 2010.
6. V. Malathy Devi, D. Chris Benner, M.A.H. Smith, A.W. Mantz, K. Sung, L.R. Brown, and A.Predoi-Cross, “Spectral line parameters including temperature dependences of self- and air-broadening in the $2 \leftarrow 0$ band of CO at 2.3 μm ,” *JQSRT* **113**, 1013-1033 (2012).
7. V. Malathy Devi, D. Chris Benner, M.A.H. Smith, A.W. Mantz, K. Sung and L.R. Brown, “Spectral line parameters including temperature dependences of air-broadening in the $2 \leftarrow 0$ bands of $^{13}\text{C}^{16}\text{O}$ and $^{12}\text{C}^{18}\text{O}$ at 2.3 μm ,” *J.Mol.Spectrosc.* **276**, 33-48 (2012).

Half-widths (air)

1. T. Nakazawa and M. Tanaka, “Measurements of Intensities and Self- and Foreign gas broadened halfwidths of spectral Lines in the CO fundamental Band,” *JQSRT* **28**, 409-416 (1982); values for transitions having $20 < |m| \leq 33$ are extrapolated, while those for $|m| > 34$ are assumed to be constant ($0.0400 \text{ cm}^{-1}/\text{atm}$).
2. Polynomial fit of several measurements (M.A.H. Smith, private communication, 2004). For details, see L.S. Rothman, D. Jacquemart, A. Barbe, D.C. Benner, M. Birk, L.R. Brown, M. Carleer, C. Chackerian Jr, K. Chance, L.H. Coudert, V. Dana, V.M. Devi, J.-M. Flaud, R.R. Gamache, A. Goldman, J.-M. Hartmann, K.W. Jucks, A.G. Maki, J.-Y. Mandin, S. Massie, J. Orphal, A. Perrin, C.P. Rinsland, M.A.H. Smith, J. Tennyson, R.N. Tolchenov, R.A. Toth, J.

Vander Auwera, P. Varanasi, and G. Wagner, "The HITRAN 2004 Molecular Spectroscopic Database," *JQSRT* **96**, 139-204 (2005).

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Half-widths (self)

1. T. Nakazawa and M. Tanaka, "Measurements of Intensities and Self- and Foreign gas broadened halfwidths of spectral Lines in the CO fundamental Band," *JQSRT* **28**, 409-416 (1982); values for transitions having $20 < |m| \leq 33$ are extrapolated, while those for $|m| > 34$ are assumed to be constant ($0.0460 \text{ cm}^{-1}/\text{atm}$).

2. Polynomial fit of several measurements (M.A.H. Smith, private communication, 2004). For details, see L.S. Rothman, D. Jacquemart, A. Barbe, D.C. Benner, M. Birk, L.R. Brown, M. Carleer, C. Chackerian Jr, K. Chance, L.H. Coudert, V. Dana, V.M. Devi, J.-M. Flaud, R.R. Gamache, A. Goldman, J.-M. Hartmann, K.W. Jucks, A.G. Maki, J.-Y. Mandin, S. Massie, J. Orphal, A. Perrin, C.P. Rinsland, M.A.H. Smith, J. Tennyson, R.N. Tolchenov, R.A. Toth, J. Vander Auwera, P. Varanasi, and G. Wagner, "The HITRAN 2004 Molecular Spectroscopic Database," *JQSRT* **96**, 139-204 (2005).

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Temperature dependence of air-broadened half-width

1. Polynomial fit of several measurements (M.A.H. Smith, private communication, 2004). For details, see L.S. Rothman, D. Jacquemart, A. Barbe, D.C. Benner, M. Birk, L.R. Brown, M. Carleer, C. Chackerian Jr, K. Chance, L.H. Coudert, V. Dana, V.M. Devi, J.-M. Flaud, R.R. Gamache, A. Goldman, J.-M. Hartmann, K.W. Jucks, A.G. Maki, J.-Y. Mandin, S. Massie, J. Orphal, A. Perrin, C.P. Rinsland, M.A.H. Smith, J. Tennyson, R.N. Tolchenov, R.A. Toth, J. Vander Auwera, P. Varanasi, and G. Wagner, "The HITRAN 2004 Molecular Spectroscopic Database," *JQSRT* **96**, 139-204 (2005).

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3. V. Malathy Devi, D. Chris Benner, M.A.H. Smith, A.W. Mantz, K. Sung, and L.R. Brown, "Spectral line parameters including temperature dependences of air-broadening in the 2←0 bands of ¹³C¹⁶O and ¹²C¹⁸O at 2.3 μm," *J.Mol.Spectrosc.* **276**, 33-48 (2012).

Pressure shift (air)

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constant (-0.0055 cm⁻¹/atm).

2. K. Sung and P. Varanasi, "Intensities, collision-broadened half-widths, and collision-induced line shifts in the second overtone band of ¹²C¹⁶O," *JQSRT* **83**, 445-458 (2004); for the 3-0 band, values for transitions having -24 ≥ m ≥ 26 are assumed to be constant (-0.0075 cm⁻¹/atm).

3. V. Malathy Devi, D. Chris Benner, M.A.H. Smith, A.W. Mantz, K. Sung, L.R. Brown, and A.Predoi-Cross, "Spectral line parameters including temperature dependences of self- and air-broadening in the 2←0 band of CO at 2.3 μm," *JQSRT* **113**, 1013-1033 (2012).

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$$\delta_{0-0}(m) = \delta_{(2-0)\text{asymmetric}} = \frac{\delta_{2-0}(m) - \delta_{2-0}(-m)}{2}.$$

6. J.-M. Hartmann and C. Boulet, "Line shape parameters for HF in a bath of argon as a test of classical path models," *J.Chem.Phys.* **113**, 9000 (2000).

$$\delta_{v'-v''}(m) = 0.5 \times (v' - v'') \times \delta_{(2-0)\text{symmetric}} + \delta_{(2-0)\text{asymmetric}}, \text{ where}$$

$$\delta_{(2-0)\text{symmetric}} = \frac{\delta_{2-0}(m) + \delta_{2-0}(-m)}{2}, \quad \delta_{(2-0)\text{asymmetric}} = \frac{\delta_{2-0}(m) - \delta_{2-0}(-m)}{2}.$$

CH₄ [6] 211, 311, 212, 312

Positions

0. The 1986 HITRAN article: Refs. Numbers 46-52 therein are needed to document the 1991 methane linelist completely.
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Half-widths (air)

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Half-widths (self)

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10. D.C. Benner, V.M. Devi, J.J. O'Brien, S. Shaji, P.T. Spickler, C.P. Houck, J.A. Coakley, J. Dolph, and K. Rankin, "Empirical line parameters of CH₄ from 10923 to 11502 cm⁻¹," *Icarus* (in preparation).

Temperature dependence of air-broadened half-width

1. L.R. Brown, D.C. Benner, J.-P. Champion, V.M. Devi, L. Fejard, R.R. Gamache, T. Gabard, J.C. Hilico, B. Lavorel, M. Loëte, G.C. Mellau, A. Nikitin, A.S. Pine, A. Predoi-Cross, C.P. Rinsland, O. Robert, R.L. Sams, M.A.H. Smith, S.A. Tashkun, and V.G. Tyuterev, "Methane Line Parameters in HITRAN," *JQSRT* **82**, 219-238 (2003).
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9. D.C. Benner, V.M. Devi, J.J. O'Brien, S. Shaji, P.T. Spickler, C.P. Houck, J.A. Coakley, J. Dolph, and K. Rankin, "Empirical line parameters of CH₄ from 10923 to 11502 cm⁻¹," *Icarus* (in preparation).

Pressure shift (air)

1. L.R. Brown, D.C. Benner, J.-P. Champion, V.M. Devi, L. Fejard, R.R. Gamache, T. Gabard,

J.C. Hilico, B. Lavorel, M. Loëte, G.C. Mellau, A. Nikitin, A.S. Pine, A. Predoi-Cross, C.P. Rinsland, O. Robert, R.L. Sams, M.A.H. Smith, S.A. Tashkun, and V.G. Tyuterev, "Methane Line Parameters in HITRAN," *JQSRT* **82**, 219-238 (2003).

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O₂ [7] 66, 68, 67

Positions

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4. Molecular Hamiltonian constants for the v"= 0 and v'= 1 of the XΣ state are from G. Rouillé et al (Ref. 1). Vibrational term values are from Krupenie (Ref. 2), upper vibrational state energy shifted to agree with band center of G. Rouillé et al for (1←0) band.
5. Molecular Hamiltonian constants for the v"= 1 and v'= 1 of the XΣ state are from G. Rouillé et al (Ref. 1). Vibrational term values are from P.H. Krupenie (Ref. 2), upper vibrational state energy shifted to agree with band center of G. Rouillé et al for (1←0) band.
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22. Molecular Hamiltonian constants for the $v''=0$ state of $X\Sigma$ are from G. Rouillé et al (Ref. 1). For the $v'=1$ of the $b\Sigma$ state, the constants of D.L. Albritton (Ref. 18) are used. Vibrational and electronic term values are from P.H. Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state.

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24. Molecular Hamiltonian constants for the $v''=1$ state of $X\Sigma$ are from G. Rouillé et al (Ref. 1). For the $v'=1$ of the $b\Sigma$ state, the constants of D.L. Albritton (Ref. 18) are used. Vibrational and electronic term values are from Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state.

25. Molecular Hamiltonian constants for the $v''=1$ state of $X\Sigma$ are from Rouillé et al. (Ref. 1). For the $v'=0$ of the $b\Sigma$ state, the constants of D.L. Albritton (Ref. 18) are used. Vibrational and electronic term values are from Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state.

26. Molecular Hamiltonian constants for the $v''=0$ state of $X\Sigma$ are from M. Mizushima and S. Yamamoto (Ref. 6). For the $v'=0$ of the $b\Sigma$ state, the constants of H. Babcock and L. Herzberg (Ref. 19) are used. Vibrational and electronic term values are from Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state. Electronic term value is shifted by $(-0.041-0.014) \text{ cm}^{-1}$, unknown reference.

27. Molecular Hamiltonian constants for the $v''=0$ state of $X\Sigma$ are from M. Mizushima and S. Yamamoto (Ref. 6). For the $v'=1$ of the $b\Sigma$ state, the constants of W.S. Benedict (Ref. 20) are used. Vibrational and electronic term values are from P.H. Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state. Electronic term value is shifted by $(-0.041-0.014) \text{ cm}^{-1}$, unknown reference.

28. Molecular Hamiltonian constants for the $v''=0$ state of $X\Sigma$ are from M. Mizushima and S. Yamamoto (Ref. 6). For the $v'=2$ of the $b\Sigma$ state, the constants from W.S. Benedict, University of Maryland (private communication) are used. Vibrational and electronic term values are from P.H. Krupenie (Ref. 2). Dunham zero point energy correction applied to zero point energy of upper vibrational state. Electronic term value is shifted by $(-0.041-0.014) \text{ cm}^{-1}$, unknown

reference.

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Intensities

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Halfwidths (self)

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Temperature dependence of air-broadened half-width

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NO₂ [10] 646

Positions

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Pressure shift (air)

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NH₃ [11] 4111, 5111

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Half-widths (air)

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Pressure shift (air)

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HNO₃ [12] 146

Positions

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Temperature dependence of air-broadened half-width

1. J.-M. Flaud, G. Brizzi, M. Carlotti, A. Perrin, and M. Ridolfi, “MIPAS database: Validation of HNO_3 line parameters using MIPAS satellite measurements,” *Atmos. Chem. Phys. Discuss.* **6**, 4251-4272 (2006).

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OH [13] 61, 81, 62

Positions

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HF [14] 19, 29

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Pressure shift (air)

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2. A.S. Pine and J.P. Looney, “N₂ and Air Broadening in the Fundamental Bands of HF and HCl,” *J.Mol.Spectrosc.* **122**, 41-55 (1987); ($\Delta v = 1$).
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HCl [15] 15, 17, 25, 27

Positions

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CIO [18] 56, 76

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H₂CO [20] 126, 136, 128

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HOCl [21] 165, 167

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N₂ [22] 44

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Halfwidths (self)

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HCN [23] 124, 134, 125

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CH₃Cl [24] 215, 217

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Intensities

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chloride in the 3.4- μm region: Line positions and intensities," *JQSRT* **112**, 2446-2462 (2011).

Half-widths (air)

- 0.** L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, "The HITRAN database: 1986 Edition," *Appl.Opt.* **26**, 4058-4097 (1987).
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Half-widths (self)

- 1.** C. Chackerian, Jr., L.R Brown, N. Lacome , and G. Tarrago, "Methyl Chloride v₅ Region Line Shape Parameters and Rotational Constants for the v₂, v₅ and 2v₃ Vibrational Bands," *J.Mol.Spectrosc.* **191**, 148-157 (1998).

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- 1.** J.-P. Bouanich, G. Blanquet, J.-C. Populaire, and J. Walrand, "N₂-Broadening for Methyl Chloride at Low Temperature by Diode-Laser Spectroscopy," *J.Mol.Spectrosc.* **208**, 72-78 (2001).

Shift

- 1.** Crude estimate of the shift based on comparison with the PNNL spectrum.

H₂O₂ [25] 1661

Positions

- 0.** L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, "The HITRAN database: 1986 Edition," *Appl.Opt.* **26**, 4058-4097 (1987).
- 1.** J.J. Hillman, D.E. Jennings, W.B. Olson, and A. Goldman, "High-Resolution Infrared Spectrum of Hydrogen Peroxide: The v₆ Fundamental Band," *J.Mol.Spectrosc.* **117**, 46-59 (1986).
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Intensities

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C₂H₂ [26] 1221, 1231

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- 14.** L. Gomez, D. Jacquemart, N. Lacome, and J.-Y. Mandin, “Line intensities of $^{12}\text{C}_2\text{H}_2$ in the 7.7 μm spectral region,” *JQSRT* **110**, 2102-2114 (2009).
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Half-widths (air)

- 0.** L.S. Rothman, R.R. Gamache, A. Goldman, L.R. Brown, R.A. Toth, H.M. Pickett, R.L. Poynter, J.-M. Flaud, C. Camy-Peyret, A. Barbe, N. Husson, C.P. Rinsland, and M.A.H. Smith, “The HITRAN database: 1986 Edition,” *Appl.Opt.* **26**, 4058-4097 (1987).
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Half-widths (self)

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Temperature dependence of air-broadened half-width

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Pressure shift (air)

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C₂H₆ [27] 1221

Positions

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Half-widths (air)

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Half-widths (self)

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Temperature dependence of air-broadened half-width

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Pressure shift (air)

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PH₃ [28] 1111

Positions

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Half-widths (air)

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COF₂ [29] 269

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SF₆ [30] 29

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Half-widths (air)

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Half-widths (self)

1. G.D.T. Tejwani and K. Fox, "Calculated self- and foreign-gas-broadened linewidths for SF₆," *JQSRT* **37**, 541-546 (1987).

Temperature dependence of air-broadened half-width

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H₂S [31] 121,141,131

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- 3.** B. Sumpf, I Meusel, and H.-D. Kronfeldt, "Self- and air-Broadening in the v₁ and v₃ bands of H₂S," *J.Mol.Spectrosc.* **177**, 143-145 (1996).
- 4.** A. Kissel, B. Sumpf, H.-D. Kronfeldt, B.A. Tikhomirov, and Yu.N. Ponomarev, "Molecular-Gas-Pressure-Induced Line-Shift and Line-Broadening in the v₂-Band of H₂S," *J.Mol.Spectrosc.* **216**, 345-354 (2002).
- 5.** B. Sumpf, A. Kissel, and H.-D. Kronfeldt, "Line-Broadening and Line-Shift in the v₁, v₃, and 2v₂ bands of H₂S," in preparation.
- 6.** Average values of Refs 2-5.

Half-widths (self)

- 1.** J. Waschull, F. Kuhnemann, and B. Sumpf, "Self-, air- and Helium Broadening of the v₂ band of H₂S," *J.Mol.Spectrosc.* **165**, 150-158 (1994).
- 2.** B. Sumpf, I Meusel, and H.-D. Kronfeldt, "Self- and air-Broadening in the v₁ and v₃ bands of H₂S," *J.Mol.Spectrosc.* **177**, 143-145 (1996).
- 3.** B. Sumpf, "Experimental Investigation of the Self-Broadening Coefficients in the v₁ + v₃ band of SO₂ and the 2v₂ band of H₂S," *J.Mol.Spectrosc.* **181**, 160-167 (1997).
- 4.** Average values of Refs 1-3.

Pressure shift (air)

- 1.** A. Kissel, B. Sumpf, H.-D. Kronfeldt, B.A. Tikhomirov, and Yu.N. Ponomarev, "Molecular-Gas-Pressure-Induced Line-Shift and Line-Broadening in the v₂-Band of H₂S," *J.Mol.Spectrosc.* **216**, 345-354 (2002).
- 2.** B. Sumpf, A. Kissel, and H.-D. Kronfeldt, "Line-Broadening and Line-Shift in the v₁, v₃, and 2v₂ bands of H₂S," in preparation
- 3.** L.S. Rothman, D. Jacquemart, A. Barbe, D.C. Benner, M. Birk, L.R. Brown, M. Carleer, C. Chackerian Jr, K. Chance, L.H. Coudert, V. Dana, V.M. Devi, J.-M. Flaud, R.R. Gamache, A.

Goldman, J.-M. Hartmann, K.W. Jucks, A.G. Maki, J.-Y. Mandin, S. Massie, J. Orphal, A. Perrin, C.P. Rinsland, M.A.H. Smith, J. Tennyson, R.N. Tolchenov, R.A. Toth, J. Vander Auwera, P. Varanasi, and G. Wagner, "The *HITRAN* 2004 Molecular Spectroscopic Database," *JQSRT* **96**, 139-204 (2005).

HCOOH [32] 126

Positions

1. A. Goldman, F.H. Murcray, D.G. Murcray, and C.P. Rinsland, "A Search for Formic Acid in the Upper Troposphere: A Tentative Identification of the 1105 cm^{-1} ν_6 band Q branch in High Resolution Balloon-borne Absorption Spectra," *Geophys.Res.Let.* **11**, 307-310 (1984); A. Goldman and J.R. Gillis, "Line Parameters and Line-by-line Calculations for Molecules of Stratospheric Interest," Progress Report, Dept. of Physics, Univ. Denver (1984).
2. A. Perrin, C.P. Rinsland, and A. Goldman, "Spectral parameters for the ν_6 region of HCOOH and its measurement in the infrared tropospheric spectrum," *J.Geophys.Res.* **104**, 18,661-18,666 (1999).
3. J. Vander Auwera, private communication (2004), based on J. Vander Auwera, "High-Resolution Investigation of the Far-Infrared Spectrum of Formic Acid," *J.Mol.Spectrosc.* **155**, 136-142 (1992).
4. A. Perrin and J. Vander Auwera, "An improved database for the $9\text{ }\mu\text{m}$ region of the formic acid spectrum," *JQSRT* **108**, 363-370 (2007).
5. A. Perrin, J. Vander Auwera, and Z. Zelinger, "High-resolution Fourier transform study of the ν_3 fundamental band of *trans*-formic acid," *JQSRT* **110**, 743-755 (2009).

Intensities

1. A. Goldman, F.H. Murcray, D.G. Murcray, and C.P. Rinsland, "A Search for Formic Acid in the Upper Troposphere: A Tentative Identification of the 1105 cm^{-1} ν_6 band Q branch in High Resolution Balloon-borne Absorption Spectra," *Geophys.Res.Let.* **11**, 307-310 (1984); A. Goldman and J.R. Gillis, "Line Parameters and Line-by-line Calculations for Molecules of Stratospheric Interest," Progress Report, Dept. of Physics, Univ. Denver (1984).
2. A. Perrin, C.P. Rinsland, and A. Goldman, "Spectral parameters for the ν_6 region of HCOOH and its measurement in the infrared tropospheric spectrum," *J.Geophys.Res.* **104**, 18,661-18,666 (1999).
3. J. Vander Auwera, private communication (2004), based on J. Vander Auwera, "High-Resolution Investigation of the Far-Infrared Spectrum of Formic Acid," *J.Mol.Spectrosc.* **155**, 136-142 (1992).
4. A. Perrin and J. Vander Auwera, "An improved database for the $9\text{ }\mu\text{m}$ region of the formic acid spectrum," *JQSRT* **108**, 363-370 (2007); J. Vander Auwera, K. Didriche, A. Perrin, and F. Keller, "Absolute line intensities for formic acid and dissociation constant of the dimer," *J.Chem.Phys* **126**, 124311 (2007).
5. A. Perrin, J. Vander Auwera, and Z. Zelinger, "High-resolution Fourier transform study of the ν_3 fundamental band of *trans*-formic acid," *JQSRT* **110**, 743-755 (2009).

Half-widths (air)

1. A. Goldman and J.R. Gillis, "Line Parameters and Line-by-line Calculations for Molecules of Stratospheric Interest," Progress Report, Dept. of Physics, Univ. Denver (1984).
2. A. Perrin and J. Vander Auwera, "An improved database for the $9\text{ }\mu\text{m}$ region of the formic acid spectrum," *JQSRT* **108**, 363-370 (2007).

Half-widths (self)

1. A. Perrin, C.P. Rinsland, and A. Goldman, "Spectral parameters for the ν_6 region of HCOOH and its measurement in the infrared tropospheric spectrum," *J.Geophys.Res.* **104**, 18,661-18,666

(1999).

2. A. Perrin and J. Vander Auwera, “An improved database for the 9 μm region of the formic acid spectrum,” *JQSRT* **108**, 363-370 (2007). Note that the value takes into account the contribution from the dimer.

Temperature dependence of air-broadened half-width

1. A. Goldman, private communication (1996).

HO₂ [33] 166

Positions

1. C. Yamada, Y. Endo, and E. Hirota, "Difference frequency laser spectroscopy of the v₁ band of the HO₂ radical," *J.Chem.Phys.* **78**, 4379-4384 (1983).
2. K. Nagai, Y. Endo, and E. Hirota, "Diode Laser Spectroscopy of the HO₂ v₂ Band," *J.Mol.Spectrosc.* **89**, 520-527 (1981).
3. D.D. Nelson, Jr., and M.S. Zahniser, "Diode Laser Spectroscopy of the v₃ Vibration of the HO₂ Radical," *J.Mol.Spectrosc.* **150**, 527-534 (1991).
4. K. Chance, K.W. Jucks, D.G. Johnson, and W.A. Traub, "The Smithsonian Astrophysical Observatory Database SAO92," *JQSRT* **52**, 447-457 (1994).

Intensities

1. M.S. Zahniser, K.E. McCurdy, and A.C. Stanton, "Quantitative Spectroscopic Studies of the HO₂ Radical: Band Strength Measurements for the v₁ and v₂ Vibrational Bands," *J.Phys.Chem.* **93**, 1065-1070 (1989).
2. K. Chance, K.W. Jucks, D.G. Johnson, and W.A. Traub, "The Smithsonian Astrophysical Observatory Database SAO92," *JQSRT* **52**, 447-457 (1994).

Half-widths (air)

1. D.D. Nelson and M.S. Zahniser "Air broadening measurements for the v₂ vibrational band of the hydroperoxyl radical," *J.Mol.Spectrosc.* **166**, 273-279 (1994).
2. K. Chance, K.W. Jucks, D.G. Johnson, and W.A. Traub, "The Smithsonian Astrophysical Observatory Database SAO92," *JQSRT* **52**, 447-457 (1994).

O [34] 6

Positions

1. L.R. Zink, K.M. Evenson, F. Matsushima, T. Nelis, and R. L. Robinson, “Atomic oxygen fine-structure 108hermosph with tunable far-infrared spectroscopy,” *Astrophys.J.* **371**, L85-L86 (1991).

Intensities

1. H.M. Pickett, R.L. Poynter, E.A. Cohen, M.L. Delitsky, J.C. Pearson, and H.S.P. Müller, “Submillimeter, Millimeter, and Microwave Spectral Line Catalog,” JPL Publication 800-23, rev. 4 (1996).

Half-widths (air)

1. Does not have the standard HITRAN definition of Lorentz air broadening, but a default value of $0.05 \text{ cm}^{-1}/\text{atm}$ was appended.

ClONO₂ [35] 5646, 7646

Positions

1. W. Bell, G. Duxbury, and D.D. Stuart, "High-Resolution Spectra of the v₄ Band of Chlorine Nitrate," *J.Mol.Spectrosc.* **152**, 283-297 (1992); A. Goldman, C.P. Rinsland, F.J. Murcray, R.D. Blatherwick, and D.G. Murcray, "High Resolution Studies of Heavy Noy Molecules in Atmospheric Spectra," *JQSRT* **52**, 367-377 (1994).

Intensities

1. A. Goldman, C.P. Rinsland, F.J. Murcray, R.D. Blatherwick, and D.G. Murcray, "High Resolution Studies of Heavy Noy Molecules in Atmospheric Spectra," *JQSRT* **52**, 367-377 (1994).

Half-widths (air)

1. A. Goldman, C.P. Rinsland, F.J. Murcray, R.D. Blatherwick, and D.G. Murcray, "High Resolution Studies of Heavy Noy Molecules in Atmospheric Spectra," *JQSRT* **52**, 367-377 (1994).

NO⁺ [36] 46

Positions

1. Positions based on a fit by D.R. Smith, AF Phillips Lab, using data of F.P. Billingsley, *Chem.Phys.Lett.* **23**, 160-166 (1973), K.P. Huber and G. Herzberg, "Molecular Spectra and Molecular Structure IV. Constants of Diatomic Molecules," Van Nostrand Reinhold Co., NY (1979), and D.R. Smith, E.R. Huppi, and R.M. Nadile, "Improved Rotational Constants for the Ground Electronic State of NO⁺ from Atmospheric Emission Spectra," private communication; D.R. Smith, E.R. Huppi, and J.O. Wise, "Observation of highly rotationally excited NO⁺ emissions in the 110hertzosphere," *J.Atmos.Solar-Terrestrial Phys.* **62**, 1189-1198 (2000).
2. Positions based on a fit by I. Gordon (2006), using data of W.C. Ho, I. Ozier, D.T. Cramb, and M.C.L. Gerry, "Diode Laser Spectroscopy of the Vibrational Fundamental of NO⁺," *J.Mol.Spectrosc.* **149**, 559-561 (1991); G. Hilpert, H. Linnartz, M. Havenith, J.J. ter Meulen, and W.L. Meerts, "Tunable infrared and far-infrared direct absorption spectroscopy of molecular ions in a supersonic jet expansion," *Chem.Phys.Letters.* **219**, 384-388 (1994); M. López-Puertas, J.-M. Flaud, J. Peralta-Calvillo, B. Funke, and S. Gil-López, "NO⁺ fundamental and first hot ro-vibrational line frequencies from MIPAS/Envisat atmospheric spectra," *J.Mol.Spectrosc.* **237**, 218-224 (2006); W.C. Bowman, E. Herbst, and F.C. De Lucia, "Millimeter and submillimeter spectrum of NO⁺," *J.Chem.Phys.* **77**, 4261-4262 (1982); E. Miescher, "Rotationsanalyse der NO⁺-banden," *Helv.Phys.Acta* **29**, 135-144 (1956).

Intensities

1. H.-J. Werner and P. Rosmus, "Ab Initio Calculations of Radiative Transition Probabilities in the X¹Σ⁺ Ground State of the NO⁺ Ion," *J.Mol.Spectrosc.* **96**, 362-367 (1982).

Half-widths (air)

1. Default value of 0.06 cm⁻¹/atm chosen, but applications are most likely not required to work in Lorentzian regime.

HOBr [37] 169, 161

Positions

1. E.A. Cohen, G.A. McRae, T.L. Tan, R.R. Friedl, J.W.C. Johns, and N. Noël, “The ν_1 Band of HOBr,” *J.Mol.Spectrosc.* **173**, 55-61 (1995).

Intensities

1. Y. Koga, H. Takeo, S. Kondo, M. Sugie, C. Matsumura, G.A. Rae, and E.A. Cohen, “The Rotational Spectra, Molecular Structure, Dipole Moment, and Hyperfine Constants of HOBr and DOBr,” *J.Mol.Spectrosc.* **138**, 467-481 (1989).

Half-widths (air)

1. A constant value of $0.06 \text{ cm}^{-1}/\text{atm}$ has been assumed for the air-broadened halfwidth with a temperature-dependence coefficient $n = 0.67$.

C₂H₄ [38] 2211, 2311

Positions

- 1.** I. Cauuet, J. Walrand, G. Blanquet, A. Valentin, L. Henry, Ch. Lambeau, M. DeVleeschouwer, and A. Fayt, "Extension to Third-Order Coriolis Terms of the Analysis of v₁₀, v₇, and v₄ Levels of Ethylene on the Basis of Fourier Transform and Diode Laser Spectra," *J.Mol.Spectrosc.* **139**, 191-214 (1990); J. Legrand, M. Azizi, F. Herlemont, and A. Fayt, "Saturation Spectroscopy of C₂H₄ Using a CO₂ Laser Sideband Spectrometer," *J.Mol.Spectrosc.* **171**, 13-21 (1995); E. Rusinek, H. Fichoux, M. Khelkhal, F. Herlemont, J. Legrand, and A. Fayt, "Subdoppler study of the v₇ band of C₂H₄ with a CO₂ Laser Sideband Spectrometer," *J.Mol.Spectrosc.* **189**, 64-73 (1998).
- 2.** A.S. Pine, "Tunable laser survey of molecular air poluants," Final Report NSF/ASRA/DAR 78-24562, MIT, Lexington, MA (1980).
- 3.** M. Rotger, V. Boudon, and J. Vander Auwera, "Line positions and intensities in the v₁₂ band of ethylene near 1450 cm⁻¹: An experimental and theoretical study," *JQSRT* **109**, 952-962 (2008).

Intensities

- 1.** I. Cauuet, J. Walrand, G. Blanquet, A. Valentin, L. Henry, Ch. Lambeau, M. DeVleeschouwer, and A. Fayt, "Extension to Third-Order Coriolis Terms of the Analysis of v₁₀, v₇, and v₄ Levels of Ethylene on the Basis of Fourier Transform and Diode Laser Spectra," *J.Mol.Spectrosc.* **139**, 191-214 (1990); W.E. Blass, L. Jennings, A.C. Ewing, S.J. Daunt, M.C. Weber, L. Senesac, S. Hager, J.J. Hillman, D.C. Reuter, and J.M. Sirota, "Absolute intensities in the v₇ band of ethylene: tunable laser measurements used to calibrate FTS broadband spectra," *JQSRT* **68**, 467-472 (2001).
- 2.** A.S. Pine, "Tunable laser survey of molecular air poluants," Final Report NSF/ASRA/DAR 78-24562, MIT, Lexington, MA (1980); M. Dang-Nhu, A.S. Pine, A. Fayt, M. DeVleeschouwer, and C. Lambeau, "Les intensités dans la pentade v₁₁, v₂ + v₁₂, 2v₁₀ + v₁₂, v₉ et v₃ + v₈ + v₁₀ de ¹²C₂H₄," *Can.J.Phys.* **61**, 514-521 (1983).
- 3.** M. Rotger, V. Boudon, and J. Vander Auwera, "Line positions and intensities in the v₁₂ band of ethylene near 1450 cm⁻¹: An experimental and theoretical study," *JQSRT* **109**, 952-962 (2008).

Half-widths (air)

- 1.** J.F. Brannon, Jr. and P. Varanasi, "Tunable Diode Laser Measurements on the 951.7393 cm⁻¹ Line of ¹²C₂H₄ at Planetary Atmospheric Temperatures," *JQSRT* **47**, 237-242 (1992).
- 2.** M. Rotger, V. Boudon, and J. Vander Auwera, "Line positions and intensities in the v₁₂ band of ethylene near 1450 cm⁻¹: An experimental and theoretical study," *JQSRT* **109**, 952-962 (2008).

Half-widths (self)

- 1.** M. Rotger, V. Boudon, and J. Vander Auwera, "Line positions and intensities in the v₁₂ band of ethylene near 1450 cm⁻¹: An experimental and theoretical study," *JQSRT* **109**, 952-962 (2008).

Temperature dependence of air-broadened half-width

- 1.** J.F. Brannon, Jr. and P. Varanasi, "Tunable Diode Laser Measurements on the 951.7393 cm⁻¹ Line of ¹²C₂H₄ at Planetary Atmospheric Temperatures," *JQSRT* **47**, 237-242 (1992).

CH₃OH [39] 2161

Positions

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, and J.W.C. Johns, “New assignments, line intensities and HITRAN database for CH₃OH at 10 μm,” *J.Mol.Spectrosc.* **228**, 453-470 (2004).
2. H.S.P. Müller, S. Thorwirth, D.A. Roth, and G. Winnewisser, “The Cologne Database for Molecular Spectroscopy, CDMS,” *A&A* **370**, L49-L52 (2001).

Intensities

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, and J.W.C. Johns, “New assignments, line intensities and HITRAN database for CH₃OH at 10 μm,” *J.Mol.Spectrosc.* **228**, 453-470 (2004).
2. H.S.P. Müller, S. Thorwirth, D.A. Roth, and G. Winnewisser, “The Cologne Database for Molecular Spectroscopy, CDMS,” *A&A* **370**, L49-L52 (2001).

Half-widths (air)

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, and J.W.C. Johns, “New assignments, line intensities and HITRAN database for CH₃OH at 10 μm,” *J.Mol.Spectrosc.* **228**, 453-470 (2004).

Half-widths (self)

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, and J.W.C. Johns, “New assignments, line intensities and HITRAN database for CH₃OH at 10 μm,” *J.Mol.Spectrosc.* **228**, 453-470 (2004).

Temperature dependence of air-broadened half-width

1. L.H. Xu, R.M. Lees, P. Wang, L.R. Brown, I. Kleiner, and J.W.C. Johns, “New assignments, line intensities and HITRAN database for CH₃OH at 10 μm,” *J.Mol.Spectrosc.* **228**, 453-470 (2004).

CH₃Br [40] 219, 211

Positions

1. D. Jacquemart, F. Kwabia Tchana, N. Lacome, and I. Kleiner, “A complete set of line parameters for CH₃Br in the 10-μm spectral region,” *JQSRT* **105**, 264-302 (2007).
2. F. Kwabia Tchana, I. Kleiner, J. Orphal, N. Lacome, and O. Bouba, “New analysis of the Coriolis-interacting v₂ and v₅ bands of CH₃⁷⁹Br and CH₃⁸¹Br,” *J Mol Spectrosc* **228**, 441-452 (2004).

Intensities

1. D. Jacquemart, F. Kwabia Tchana, N. Lacome, and I. Kleiner, “A complete set of line parameters for CH₃Br in the 10-μm spectral region,” *JQSRT* **105**, 264-302 (2007).
2. F. Kwabia Tchana, D. Jacquemart, N. Lacome, I. Kleiner, and J. Orphal, “Absolute line intensities in methyl bromide: The 7-μm region,” *J Mol Spectrosc* **235**, 132-143 (2006).

Halfwidths (air)

1. D. Jacquemart, F. Kwabia Tchana, N. Lacome, and I. Kleiner, “A complete set of line parameters for CH₃Br in the 10-μm spectral region,” *JQSRT* **105**, 264-302 (2007).

Halfwidths (self)

1. D. Jacquemart, F. Kwabia Tchana, N. Lacome, and I. Kleiner, “A complete set of line parameters for CH₃Br in the 10-μm spectral region,” *JQSRT* **105**, 264-302 (2007).

Temperature dependence of air-broadened half-width

1. D. Jacquemart and H. Tran, “Temperature dependence of self- and N₂-broadening coefficients for CH₃Br in the 10-μm spectral region,” *JQSRT* **109**, 569-579 (2008).

CH₃CN [41] 2124

Positions

1. C.P. Rinsland, V. Malathy Devi, D. Chris Benner, T.A. Blake, R.L. Sams, L.R. Brown, I. Kleiner, A. Dehayem-Kamadjeu, H.S.P. Müller, R.R. Gamache, D.L. Niles, and T. Masiello, “Multispectrum analysis of the v₄ band of CH₃CN: Positions, intensities, self- and N₂-broadening, and pressure-induced shifts,” *JQSRT* **109**, 974-994 (2008).

Intensities

1. C.P. Rinsland, V. Malathy Devi, D. Chris Benner, T.A. Blake, R.L. Sams, L.R. Brown, I. Kleiner, A. Dehayem-Kamadjeu, H.S.P. Müller, R.R. Gamache, D.L. Niles, and T. Masiello, “Multispectrum analysis of the v₄ band of CH₃CN: Positions, intensities, self- and N₂-broadening, and pressure-induced shifts,” *JQSRT* **109**, 974-994 (2008).

Halfwidths (air)

1. C.P. Rinsland, V. Malathy Devi, D. Chris Benner, T.A. Blake, R.L. Sams, L.R. Brown, I. Kleiner, A. Dehayem-Kamadjeu, H.S.P. Müller, R.R. Gamache, D.L. Niles, and T. Masiello, “Multispectrum analysis of the v₄ band of CH₃CN: Positions, intensities, self- and N₂-broadening, and pressure-induced shifts,” *JQSRT* **109**, 974-994 (2008).

Halfwidths (self)

1. C.P. Rinsland, V. Malathy Devi, D. Chris Benner, T.A. Blake, R.L. Sams, L.R. Brown, I. Kleiner, A. Dehayem-Kamadjeu, H.S.P. Müller, R.R. Gamache, D.L. Niles, and T. Masiello, “Multispectrum analysis of the v₄ band of CH₃CN: Positions, intensities, self- and N₂-broadening, and pressure-induced shifts,” *JQSRT* **109**, 974-994 (2008).

Temperature dependence of air-broadened half-width

1. C.P. Rinsland, V. Malathy Devi, D. Chris Benner, T.A. Blake, R.L. Sams, L.R. Brown, I. Kleiner, A. Dehayem-Kamadjeu, H.S.P. Müller, R.R. Gamache, D.L. Niles, and T. Masiello, “Multispectrum analysis of the v₄ band of CH₃CN: Positions, intensities, self- and N₂-broadening, and pressure-induced shifts,” *JQSRT* **109**, 974-994 (2008).

Shifts

1. C.P. Rinsland, V. Malathy Devi, D. Chris Benner, T.A. Blake, R.L. Sams, L.R. Brown, I. Kleiner, A. Dehayem-Kamadjeu, H.S.P. Müller, R.R. Gamache, D.L. Niles, and T. Masiello, “Multispectrum analysis of the v₄ band of CH₃CN: Positions, intensities, self- and N₂-broadening, and pressure-induced shifts,” *JQSRT* **109**, 974-994 (2008).

CF₄ [42] 29

Positions

1. V. Boudon, Université de Bourgogne, private communication (2008).

Intensities

1. V. Boudon, Université de Bourgogne, private communication (2008).

Half-widths (air)

1. S. Höjer and R.D. May, “Air-Broadening Coefficients for the v₃ Band of CF₄,” *J.Mol.Spectrosc.* **178**, 139-142 (1996).

Half-widths (self)

1. Estimate (0.08 cm⁻¹atm⁻¹).

Temperature dependence of air-broadened half-width

1. S. Höjer and R.D. May, “Air-Broadening Coefficients for the v₃ Band of CF₄,” *J.Mol.Spectrosc.* **178**, 139-142 (1996).

C₄H₂ [43] 2211

Positions

1. L. Bizzocchi,C. Degli Esposti, and L. Dore. “Submillimetre-wave spectrum of diacetylene and diacetylene-d2,” *Mol.Phys.* **108**, 2315-2323 (2010).
2. L. Bizzocchi, F. Tamassia, C.D. Esposti, L. Fusina, E Canè, and L. Dore, “High-resolution infrared spectroscopy of diacetylene below 1000 cm⁻¹,” *Mol.Phys.* **109**, 2181-2190 (2011).
3. A. Jolly, A. Fayt, Y. Benilan, D. Jacquemart, C.A. Nixon, and D.E. Jennings, “The v₈ bending mode of diacetylene: from laboratory spectroscopy to the detection of ¹³C isotopologues in Titan’s atmosphere,” *Ap.J.* **714**, 852-859 (2010).

Intensities

1. K. Matsumura and T. Tanaka, “Microwave vibration rotation spectrum of diacetylene,” *J.Mol.Specr.* **96**, 219-233(1982). (Integrated band intensity); L. Bizzocchi,C.D. Esposti, and L. Dore. “Submillimetre-wave spectrum of diacetylene and diacetylene-d2,” *Mol.Phys.* **108**, 2315-2323 (2010). (Relative rotational line intensities).
2. L. Bizzocchi, F. Tamassia, C.D. Esposti, L. Fusina, E Canè and L. Dore, “High-resolution infrared spectroscopy of diacetylene below 1000 cm⁻¹,” *Mol.Phys.* **109**, 2181-2190 (2011). (Relative rotational line intensities). Integrated band intensities are taken from Ref. [3].
3. A. Jolly, A. Fayt, Y. Benilan, D. Jacquemart, C.A. Nixon, and D.E. Jennings, “The v₈ bending mode of diacetylene: from laboratory spectroscopy to the detection of ¹³C isotopologues in Titan’s atmosphere,” *Ap.J.* **714**, 852-859 (2010).

HC₃N [44] 1224

Positions

1. H.S.P. Müller, F. Schloder, J. Stutzki, and G. Winnewisser, “The Cologne Database for Molecular Spectroscopy, CDMS: a useful tool for astronomers and spectroscopists,” *J.Mol.Struct.* **742**, 215-227 (2005).
2. A. Jolly, Y. Benilan, and A. Fayt, “New infrared integrated band intensities for HC₃N and extensive line list for the v₅ and v₆ bending modes,” *J.Mol.Spectrosc.* **242**, 46-54 (2007).

Intensities

1. H.S.P. Müller, F. Schloder, J. Stutzki, and G. Winnewisser, “The Cologne Database for Molecular Spectroscopy, CDMS: a useful tool for astronomers and spectroscopists,” *J.Mol.Struct.* **742**, 215-227 (2005).
2. A. Jolly, Y. Benilan, and A. Fayt, “New infrared integrated band intensities for HC₃N and extensive line list for the v₅ and v₆ bending modes,” *J.Mol.Spectrosc.* **242**, 46-54 (2007).

Halfwidths (air)

1. A. Jolly, Y. Benilan, and A. Fayt, “New infrared integrated band intensities for HC₃N and extensive line list for the v₅ and v₆ bending modes,” *J.Mol.Spectrosc.* **242**, 46-54 (2007).

Halfwidths (self)

1. A. Jolly, Y. Benilan, and A. Fayt, “New infrared integrated band intensities for HC₃N and extensive line list for the v₅ and v₆ bending modes,” *J.Mol.Spectrosc.* **242**, 46-54 (2007).

Temperature dependence of air-broadened half-width

1. A. Jolly, Y. Benilan, and A. Fayt, “New infrared integrated band intensities for HC₃N and extensive line list for the v₅ and v₆ bending modes,” *J.Mol.Spectrosc.* **242**, 46-54 (2007).

H₂ [45] 11, 12

Positions

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Intensities

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CS [46] 22, 23, 24, 32

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Intensities

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Half-widths (air)

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Half-widths (self)

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Temperature dependence of air-broadened half-width

1. Default value of 0.75 chosen.

SO₃ [47] 26

Positions

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Molecules and isotopologues in line-by-line portion of HITRAN 2012 edition

Molecule Number	Molecule	Isotopologue (AFCGL notation)	Fractional Abundance	Spectral Coverage (cm ⁻¹)	Number of lines	Total number
1	H ₂ O	161	0.9973	0 – 25711	142 045	
		181	1.999 10 ⁻³	0 – 19918	39 903	
		171	3.719 10 ⁻⁴	0 – 19946	27 544	
		162	3.107 10 ⁻⁴	0 – 22708	13 237	
		182	6.230 10 ⁻⁷	0 – 3825	1611	
		172	1.158 10 ⁻⁷	1234 – 1599	175	224 515
2	CO ₂	626	0.9842	345 – 12785	169 292	
		636	1.106 10 ⁻²	406 – 12463	70 611	
		628	3.947 10 ⁻³	0 – 9558	116 482	
		627	7.339 10 ⁻⁴	0 – 9600	72 525	
		638	4.434 10 ⁻⁵	489 – 6745	26737	
		637	8.246 10 ⁻⁶	583 – 6769	2953	
		828	3.957 10 ⁻⁶	491 – 8161	7118	
		827	1.472 10 ⁻⁶	626 – 5047	821	
		727	1.368 10 ⁻⁷	535 – 5933	5187	
		838	4.446 10 ⁻⁸	4599 – 4888	121	
3	O ₃	666	0.9929	0 – 6997	261 886	
		668	3.982 10 ⁻³	0 – 2768	44 302	
		686	1.991 10 ⁻³	1 – 2740	18 887	
		667	7.405 10 ⁻⁴	0 – 2122	65 106	
		676	3.702 10 ⁻⁴	0 – 2101	31 935	
4	N ₂ O	446	0.9903	0 – 7797	33 074	
		456	3.641 10 ⁻³	5 – 5086	4 222	
		546	3.641 10 ⁻³	4 – 4704	4 592	
		448	1.986 10 ⁻³	542 – 4672	4 250	
		447	3.693 10 ⁻⁴	550 – 4430	1 705	47 843
5	CO	26	0.9865	3 – 8465	1 019	
		36	1.108 10 ⁻²	3 – 6274	797	
		28	1.978 10 ⁻³	3 – 6267	770	
		27	3.679 10 ⁻⁴	3 – 6339	728	
		38	2.222 10 ⁻⁵	3 – 6124	712	
		37	4.133 10 ⁻⁶	1807 – 6197	580	4 606
6	CH ₄	211	0.9883	0 – 11502	336 830	
		311	1.110 10 ⁻²	0 – 11319	72 420	
		212	6.158 10 ⁻⁴	7 – 6511	54 550	
		312	6.918 10 ⁻⁶	959 – 1695	4 213	468 013
7	O ₂	66	0.9953	0 – 15928	1 787	
		68	3.991 10 ⁻³	1 – 15853	875	
		67	7.422 10 ⁻⁴	0 – 14537	11 313	13 975

Molecule Number	Molecule	Isotopologue (AFGL notation)	Fractional Abundance	Spectral Coverage (cm ⁻¹)	Number of lines	Total number
8	NO	46	0.9940	0 – 9274	103 701	
		56	3.654 10 ⁻³	1609 – 2061	699	105 079
		48	1.993 10 ⁻³	1602 – 2039	679	
9	SO ₂	626	0.9457	0 – 4093	72 460	
		646	4.195 10 ⁻²	0 – 2501	22 661	95 121
10	NO ₂	646	0.9916	0 – 3075	104 223	104 223
11	NH ₃	4111	0.9959	0 – 7000	45 302	
		5111	3.661 10 ⁻³	0 – 5180	1 090	46 392
12	HNO ₃	146	0.9891	0 – 1770	903 854	
		156	3.636 10 ⁻³	0 – 923	58 108	961 962
13	OH	61	0.9975	0 – 19268	30 772	
		81	2.000 10 ⁻³	0 – 329	295	31 979
		62	1.554 10 ⁻⁴	0 – 332	912	
14	HF	19	0.9998	24 – 46985	10 073	
		29	1.557 10 ⁻⁴	13 – 47365	24 303	34 376
15	HCl	15	0.7576	8 – 13459	11 879	
		17	0.2422	8 – 10995	11 907	
		25	1.180 10 ⁻⁴	5 – 33284	29 994	83 691
		27	3.774 10 ⁻⁵	5 – 33258	29 911	
16	HBr	19	0.5068	13 – 16034	3 039	
		11	0.4931	13 – 16032	3 031	
		29	7.894 10 ⁻⁵	7 – 8781	1 455	8 980
		21	7.680 10 ⁻⁵	7 – 8778	1 455	
17	HI	17	0.9998	10 – 13908	3 161	
		27	1.557 10 ⁻⁴	5 – 7625	1 590	4 751
18	ClO	56	0.7559	0 – 1208	5 721	
		76	0.2417	0 – 1200	5 780	11 501
19	OCS	622	0.9374	0 – 4200	15 618	
		624	4.158 10 ⁻²	0 – 4166	6 087	
		632	1.053 10 ⁻²	0 – 4056	3 129	29 361
		623	7.399 10 ⁻³	0 – 4164	2 886	
		822	1.880 10 ⁻³	0 – 4046	1 641	
20	H ₂ CO	126	0.9862	0 – 3100	40 670	
		136	1.108 10 ⁻²	0 – 117	2 309	44 601
		128	1.978 10 ⁻³	0 – 191	1 622	
21	HOCl	165	0.7558	1 – 3800	8 877	
		167	0.2417	1 – 3800	7 399	16 276
22	N ₂	44	0.9927	11 – 9355	1 107	
		45	7.478 10 ⁻³	11 – 2578	161	1 268

Molecule Number	Molecule	Isotologue (AFGL notation)	Fractional Abundance	Spectral Coverage (cm ⁻¹)	Number of lines	Total number
23	HCN	124 134 125	0.9851 $1.107 \cdot 10^{-2}$ $3.622 \cdot 10^{-3}$	0 – 3424 2 – 3405 2 – 3420	2 955 652 646	4 253
24	CH ₃ Cl	215 217	0.7489 0.2395	0 – 3198 0 – 3198	107 642 104 854	212 496
25	H ₂ O ₂	1661	0.9950	0 – 1731	126 983	126 983
26	C ₂ H ₂	1221 1231 1222	0.9776 $2.197 \cdot 10^{-2}$ $3.046 \cdot 10^{-4}$	604 – 9890 613 – 6589 1 – 789	12 613 285 7 512	20 410
27	C ₂ H ₆	1221 1231	0.9770 $2.195 \cdot 10^{-2}$	706 – 3001 725 – 919	43 592 6 037	49 629
28	PH ₃	1111	0.9995	0 – 3602	22 189	22 189
29	COF ₂	269 369	0.9865 $1.108 \cdot 10^{-2}$	696 – 2002 686 – 815	168 793 15 311	184 104
30	SF ₆	29	0.9502	580 – 996	2 889 065	2 889 065
31	H ₂ S	121 141 131	0.9499 $4.214 \cdot 10^{-2}$ $7.498 \cdot 10^{-3}$	2 – 11330 5 – 11227 5 – 11072	36 561 11 352 6 322	54 325
32	HCOOH	126	0.9839	10 – 1890	62 684	62 684
33	HO ₂	166	0.9951	0 – 3676	38 804	38 804
34	O	6	0.9976	68 – 159	2	2
35	ClONO ₂	5646 7646	0.7496 0.2397	763 – 798 765 – 791	21 988 10 211	32 199
36	NO ⁺	46	0.9940	1634 – 2531	1 206	1 206
37	HOBr	169 161	0.5056 0.4919	0 – 316 0 – 316	2 177 2 181	4 358
38	C ₂ H ₄	221 231	0.9773 $2.196 \cdot 10^{-2}$	701 – 3243 2947 – 3181	18 097 281	18 378
39	CH ₃ OH	2161	0.9859	0 – 1408	19 897	19 897
40	CH ₃ Br	219 211	0.5010 0.4874	794 – 1706 796 – 1697	18 692 18 219	36 911
41	CH ₃ CN	2124	0.9739	890 – 946	3 572	3 572
42	CF ₄	29	0.9889	594 – 1313	60 033	60 033
43	C ₄ H ₂	2211	0.9560	0 – 758	124 126	124 126
44	HC ₃ N	1224	0.9633	0 – 760	180 332	180 332

45	H ₂	11	0.9997	15 – 36024	4 017	9 146
		12	$3.114 \cdot 10^{-4}$	3 – 36406	5 129	
46	CS	22	0.9396	1 – 2586	1 088	
		24	$4.168 \cdot 10^{-2}$	1 – 1359	396	
		32	$1.056 \cdot 10^{-2}$	1 – 1331	396	2 078
		23	$7.417 \cdot 10^{-3}$	1 – 156	198	
47	SO ₃	26	0.9434	0 – 2825	10 881	10 881

Note: Rows highlighted in pink are for molecules that have been relegated to a sub-folder since they do not have sufficient hot bands included.

Uncertainty Codes used in HITRAN Database

Line position and Pressure shift (cm ⁻¹)		Intensity, Halfwidths, and Temperature-dependence	
Code	Uncertainty Range	Code	Uncertainty Range
0	≥ 1 , or Unreported	0	Unreported or Unavailable
1	≥ 0.1 and < 1 .	1	Default or Constant
2	≥ 0.01 and < 0.1	2	Average or Estimate
3	≥ 0.001 and < 0.01	3	$\geq 20\%$
4	≥ 0.0001 and < 0.001	4	$\geq 10\%$ and $< 20\%$
5	≥ 0.00001 and < 0.0001	5	$\geq 5\%$ and $< 10\%$
6	≥ 0.000001 and < 0.00001	6	$\geq 2\%$ and $< 5\%$
7	≥ 0.0000001 and < 0.000001	7	$\geq 1\%$ and $< 2\%$
8	≥ 0.00000001 and < 0.0000001	8	$< 1\%$
9	Better than 0.00000001		